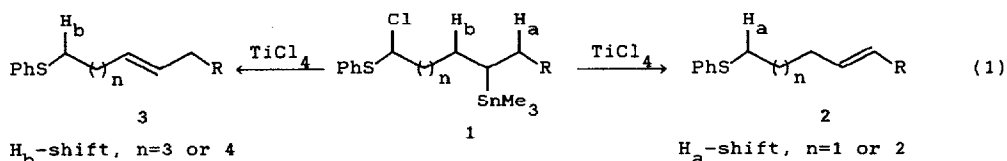


CHIRALITY TRANSFER IN THE 1,5-HYDRIDE SHIFT OF A STANNYL COMPOUND
 HAVING CATIONIC CENTER AT δ -POSITION

Tadashi Sato,* Masayuki Haramura, and Naoki Taka
 Department of Applied Chemistry, Waseda University,
 Ookubo 3, Shinjuku-ku, Tokyo 169, Japan

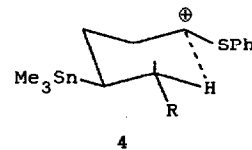
Chirality transfer with 7 : 1 selectivity was realized in the $I_2/AgOTf$ -induced 1,5-hydride shift of 2-methyl-5(R)-trimethylstannyl-1-hexene to 6-iodo-5(S)-methyl-1-hexene.

Some time ago, we reported¹⁾ that the treatment of 1 with $TiCl_4$ induced an elimination of chloride anion with concomitant hydride shift, resulting in the olefin formation (eq 1). Notably, only H_a -shift was observed with the substrates of $n = 1$ or 2, producing 2, while only H_b -shift was observed when $n = 3$ or 4, producing 3. The reaction was also noted for its high trans selectivity.

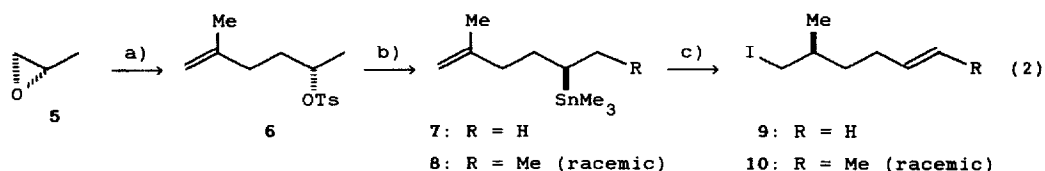


These facts induced us to propose a reaction scheme involving a cyclic transition state 4, in which the trimethylstannyl group and migrating hydrogen atom occupy the anti-periplanar position.²⁾

With a view to prove the scheme, as well as to extend the scope of the reaction, we envisioned to achieve a chirality transfer from the tin-bearing carbon to the cationic center. For this purpose, the creation of a prochiral cationic center is requisite, and we manipulated a reaction system shown in eq 2.



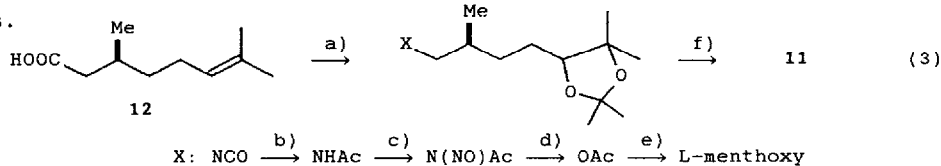
The optically active 7 was prepared from (S)-methyloxirane 5 through a reaction sequence shown in eq 2. We assigned (R)-configuration for 7 because the reaction of a tosylate with Me_3SnLi is known to proceed with



a) i) 2-methyl-2-propenylmagnesium bromide, ii) $TsCl/Py$; b) Me_3SnLi ; c) $I_2/AgOTf$;

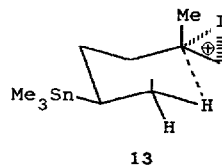
complete inversion.³⁾ When treated with $I_2/AgOTf$ in CH_2Cl_2 at $-78^\circ C$, **7** underwent the 1,5-hydride shift to afford terminal olefin **9** in about 70% yield. In order to evaluate the optical purity, the iodide **9** was converted to a diastereomeric mixture of the corresponding L-menthyl ether **11**, and the diastereomeric ratio was determined on a capillary gas chromatography. Two peaks with area ratio of 7 to 1 were observed, while the ratio was 1 to 1 in case of the menthyl ether derived from racemic **7**.

The absolute stereochemistry of the newly-created chiral center of the major product was assigned as (S), since the product was identical with the authentic **11**, prepared from (S)-citronellic acid **12** (eq 3), on GLC and NMR analyses.



- a) i) $H_2O_2/HCOOH$, $40^\circ C$; ii) acetone/ $ZnCl_2$; iii) $ClCOEt/Et_3N/NaNO_3/acetone$, $-5 \sim 0^\circ C$; iv) reflux/toluene; b) reflux/ $AcOH$; c) $NaNO_2/AcOH-Ac_2O$, $0^\circ C$; d) reflux/hexane; e) i) $NaOH/EtOH-H_2O$, reflux; ii) $TsCl/Py$, $0^\circ C$; iii) L-menthol/ $Na/toluene$, reflux; f) i) $HCl/EtOH-H_2O$, rt; ii) $NaIO_4/dioxane-H_2O$, rt; iii) $Ph_3P(Br)Me/n-BuLi/Et_2O$, rt.

The results show that the reaction proceeds through a cyclic transition state **13**, in which trimethylstannyl and iodonium π -complex moieties occupy quasi-equatorial positions. The introduction of a chiral center at a position remote from any chiral functionality is one of the challenging problems in organic synthesis. Although the remote chirality transfer is observed in sigmatropic rearrangement, only few cases⁴⁾ are known with non-conjugated system.



The I^+ -induced 1,5-hydride shift was also attained with (\pm)-**8**, which gave (\pm)-**10** in 71% yield. The typical procedure is as follows. To a mixture of $AgOTf$ (0.5 mmol) and **7** or **8** (0.4 mmol) in CH_2Cl_2 (4 ml), was added a solution of iodine (0.5 mmol) in CH_2Cl_2 (4 ml) at $-78^\circ C$ under nitrogen, and the mixture was stirred for 2 h at this temperature. The crude material, after washing with $NaHCO_3$ aq, and treating with a short column of silica gel, gave the product as an almost pure sample.

Financial support from the Asahi Glass Foundation is acknowledged.

References

- 1) E. Murayama, M. Uematsu, H. Nishio, and T. Sato, *Tetrahedron Lett.*, **25**, 313 (1984).
- 2) I. Fleming and M. Rowley, *Tetrahedron*, **42**, 3181 (1986).
- 3) J. S. Filippo Jr. and J. Silbermann, *J. Am. Chem. Soc.*, **104**, 2831 (1982); E. C. Ashby, *Acc. Chem. Res.*, **21**, 414 (1988).
- 4) P. A. Bartlett, *Tetrahedron*, **36**, 2 (1980).

(Received in Japan 26 June 1989)